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Protease-Activatable Nanozyme with Photoacoustic and Tumor-Enhanced Magnetic Resonance Imaging for Photothermal Ferroptosis Cancer Therapy

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Despite the promise of ferrotherapy in cancer treatment, current ferrous therapeutics suffer from compromised antitumor ferroptosis efficacy and low specificity for tumors. Herein, a protease-activatable nanozyme (Fe₃O₄@Cu_{1 77}Se) is reported for photoacoustic and tumor-enhanced magnetic resonance imaging (MRI)-guided second near-IR photothermal ferroptosis cancer therapy. Fe₃O₄@Cu₁₇₇Se remains stable in physiological conditions, but disintegrates to increase reactive intratumoral ferrous supply for elevated hydroxyl radical generation by Fenton reaction and GSH depletion in response to overexpressed matrix metalloproteinases in tumor microenvironment, leading to amplified ferroptosis of tumor cells as well as enhanced T₂-weighted MRI contrast. Further integration with second near-IR photoirradiation to generate localized heat not only triggers effective photothermal therapy and photoacoustic imaging but more importantly, potentiates Fenton reaction to promote ferroptotic tumor cell death. Such synergism leads to the polarization of tumor-associated macrophage from the tumor-promoting M2 type to the tumor-killing M1 type, and induces the immunogenic cells death of tumor cells, which in turn promotes the maturation of dendritic cells and infiltration of cytotoxic T lymphocytes in tumor, contributing to significant tumor suppression. This study presents a novel activatable ferrous nanotheranostics for spatial-temporal control over antitumor ferroptosis responses.

conversion of endogenous hydrogen peroxide (H₂O₂) into highly cytotoxic hydroxyl radical (•OH) to induce a lethal level of lipid peroxidation (LPO).^[1] Distinct from apoptosis, necrosis, and autophagy, ferroptosis is not affected by the common tumorassociated mutations such as upregulation of anti-apoptotic proteins or downregulation of pro-apoptotic proteins and thus, offers a promising anti-tumor therapeutic strategy. Considering the vital role of iron in ferroptosis induction, a variety of smallmolecule or iron-based nanotherapeutics, such as ferumoxytol,^[2] amorphous iron nanoparticles,^[3] inorganic iron nanoparticles,^[4] and iron-organic frameworks,^[5] has been developed. Unfortunately, the unfavorable tumor microenvironment (TME) compromises their catalytic activities, and thus, are often required a very high dosage of iron (75 mg kg⁻¹ body weight by intravenous injection) to achieve desirable tumor inhibition.^[6] The nonspecific biodistribution and "always-on" pharmacological activities of iron-based therapeutic agents would inevitably incur the "off-

1. Introduction

Ferroptosis refers to a new non-apoptotic programmed cell death pathway that mainly relies on iron-dependent catalytic

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target" toxicity, which may cause collateral side effects toward normal tissues. To mitigate the damage to normal tissues and boost the therapeutic efficacy, current efforts have been devoted to designing stimuli-responsive nano-therapeutics, of which

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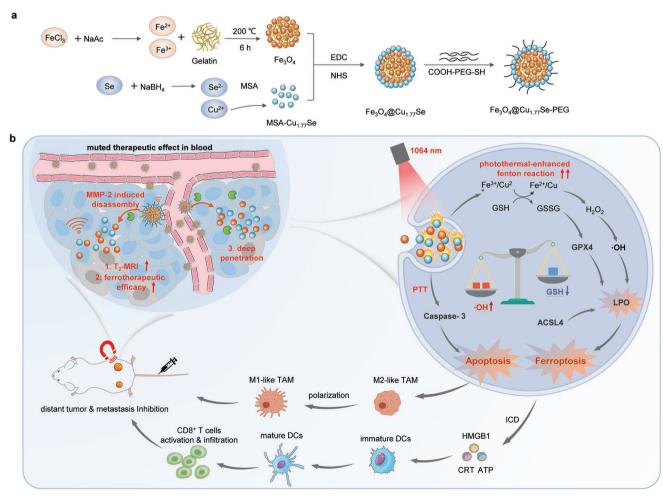
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the therapeutic abilities could be "turned on" in response to specific stimuli in tumor or within TME (e.g., low pH, highly expressed enzyme, hypoxia, etc.).^[7] These nanoplatforms have shown tumor-specific therapeutic activation with reduced toxicity, which are however less investigated in tumor-specific ferrotherapy.

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To enhance anti-tumor ferrotherapeutic performance, elevated temperature in the presence of photothermal conversion materials was reported to accelerate the Fenton reaction kinetics.^[8] On the one hand, the photoirradiation facilities the reactive oxygen species (ROS) generation of iron-therapeutics through direct electron transfer and photo-enhanced Fenton reaction, and concurrently, the ROS generation can be improved by the photothermal effect.^[9] Kinetics studies indicate that the Fenton reaction rate was enhanced by up to four folds upon increasing the temperature from 20 to 50 °C.^[6] On the other hand, hyperthermia affects intracellular enzymatic activity and breaks the inherent resistance (e.g., ferroptosis-related proteins, labile iron pool, and respiratory enzymes) of cancer cells, and thus, generating desirable anti-tumor therapeutic effects. Compared to the conventional first near-IR (NIR-I, 700-1000 nm) window, second near-IR (NIR-II, 1000-1700 nm) window exhibits significantly reduced tissue attenuation, deep tissue penetration along with elevated maximum permissible energy exposure to skin of light,^[10] and thus, has inspired researchers to constantly explore NIR-II photothermal materials including gold nanoparticles,^[11] carbon materials,^[12] copper sulfide-based nanoparticles,^[13] and organic polymer^[14] through precise particle size/morphology control, delicate surface modification or component selection. Considerable investigations were conducted for photothermal-enhanced Fenton nanoagents,[8a] which only a few examples were reported for in vivo NIR-II photothermal ferrotherapy, including transformable semiconducting nanozyme,^[15] single-atom Pd nanozyme,^[16] cysteineresponsive charge-transfer complex nanoparticles.^[17] However, modulating the TME-activatable nanozyme in hybrid cancer NIR-II ferrotherapy has been underexploited so far. Moreover, the mechanistic investigation responsible for the effective inhibition of tumor and metastasis remains elusive.

Herein, we report a nanozyme ($Fe_3O_4@Cu_{1.77}Se$) with protease-activatable, NIR-II photothermal-enhanced catalytic and GSH depletion activities for photoacoustic (PA) and magnetic resonance imaging (MRI)-guided photothermal ferroptosis cancer therapy (**Scheme 1**). Gelatin, a polypeptide that can be degraded by matrix metalloproteinases (MMPs), such as MMP-2 or MMP-9, overexpressed in TME, serves as template



Scheme 1. a) Synthesis and b) biological mechanism of MMP-2-activatable $Fe_3O_4@Cu_{1.77}$ Se nanozymes with enhanced MR-imaging for second near-IR photothermal ferroptosis cancer therapy.



for the assembly of superparamagnetic Fe₃O₄ nanoparticles into Fe₃O₄ nanoclusters, followed by the conjugation of Cu₁₇₇Se via EDC/NHS chemistry (Scheme 1a). The nanozymes remain intact followed by systematic circulation and their magneticresponsive behavior allows the magnetic field-guided delivery to tumor site. Upon reaching the tumor site, the overexpressed MMPs in the TME disassemble the Fe₃O₄@Cu₁₇₇Se, not only transforming them from 124.2 nm into small fragments (≈14.6 ± 6.9 nm), favoring deep penetration into solid tumor, but also enhance intratumoral iron pool to elevate the T₂-weighted MRI contrast and the ferrotherapeutic effect (Scheme 1b). Further NIR-II photoirradiation to induce localized heat not only triggers effective PTT to enhance the apoptosis but also significantly potentiates Fenton reaction to generate hydroxyl radical and deplete GSH to induce more lipid peroxidation (LPO) for enhanced ferroptosis. Such synergism polarizes M2 tumor-associated macrophages to M1 tumor-associated macrophages, and also, induces immunogenic cell death to enhance the recruitment and infiltration of cytotoxic T cells into the immunogenetic cold tumor (4T1). Therefore, it not only leads to the elimination of primary tumor, but also elicits systemic anti-tumor immunity, contributing to regression of non-treated distant tumors and inhibition of lung metastasis. The combinational on-site and precise spatial-temporal activation strategy triggers the therapeutic process specifically within tumor tissues rather than normal ones, resulting in the concurrent high photothermal ferroptosis cancer therapeutic efficacy and negligible damage to normal tissues.

2. Results and Discussion

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The nanozyme, Fe₃O₄@Cu₁₇₇Se, was prepared via a stepby-step conjugation. Briefly, Fe₃O₄ nanoclusters, assembled from superparamagnetic Fe₃O₄ nanoparticles with gelatin as a mediator, serve as seeds, followed by conjugation of Cu₁₇₇Se via seed-mediated growth. Transmission electron microscopy (TEM) images showed that Fe₃O₄, Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se were uniform, spherical and monodisperse, with the diameters determined to be 116.2 ± 7.1 , 4.1 ± 1.3 , and 124.2 ± 3.0 nm, respectively (Figure 1a; Figures S1, S2, Supporting Information). The lattice distance of 0.21 and 0.33 nm in the highresolution TEM (HRTEM) images of Fe₃O₄@Cu_{1.77}Se correlated with the (400) and (111) plane of Fe_3O_4 and $Cu_{1.77}Se$, respectively (Figure 1b). Elemental mappings revealed the presence and homogeneous distribution of Fe, O, Cu, and Se elements in the entire architecture of Fe₃O₄@Cu_{1.77}Se (Figure 1c). X-ray photoelectron spectroscopy (XPS) further confirmed the presence of Cu, Fe, and Se in Fe₃O₄@Cu_{1.77}Se (Figures 1d,e; Figure S3, Supporting Information). Mixed valences of Cu²⁺ (binding energy at 933.6 and 954.1 eV) and Cu+ (binding energy at 932.2 and 952.1 eV) were assigned to the peaks of Cu $2p_{3/2}$ and Cu $2p_{1/2}$, respectively. The content of Cu²⁺ and Cu⁺ was calculated to be 12.9% and 87.1%, respectively. The peaks centered at 709.6 and 719.3 eV in the Fe $2p_{3/2}$ spectrum can be assigned to Fe²⁺ and Fe³⁺, respectively. The content of Fe²⁺ and Fe³⁺ in the nanozymes was determined to be 50.5% and 49.5%, respectively. Consistently, the molar ratio of Fe/Cu in the nanozymes, characterized by the inductively coupled plasma optical emission spectrometer (ICP-OES), was determined to be ~1:0.83 (Table S1, Supporting Information). X-ray diffraction (XRD) patterns revealed the presence of Fe₃O₄ (JCPDS 19–0629) and Cu_{2-x}Se (JCPDS 06–0680) peaks in Fe₃O₄@Cu_{1.77}Se (Figure 1f). Magnetic hysteresis loops indicated that Fe₃O₄ and Fe₃O₄@Cu_{1.77}Se were superparamagnetic at room temperature, with the saturation magnetization values of 71.9 and 44.6 emu g⁻¹, respectively (Figure 1g).

Next, the photothermal properties of Fe₃O₄@Cu₁₇₇Se were investigated. Fe₃O₄ had very low NIR-II absorption, regardless of a broad absorption from visible to NIR region (Figure S4, Supporting Information). After conjugation with Cu177Se, a significantly enhanced absorption peak from visible to NIR-II range, especially from 1000 to 1400 nm, was observed, indicating the potential of Fe₃O₄@Cu₁₇₇Se as NIR-II photothermal agents. Therefore, upon 1064 nm photoirradiation, Fe₃O₄@Cu₁₇₇Se exhibited power-dependent temperature increase (Figure 1h; Figure S5, Supporting Information). The maximum temperature reached ≈53 °C at Fe₃O₄@Cu_{1.77}Se concentration of 100 μ g mL⁻¹ in 10 min (0.75 W cm⁻²), whereas water showed negligible temperature increase (Figure S6, Supporting Information). The photothermal conversion efficacy (η) was determined to be 67.6% (Figure 1i). Moreover, Fe₃O₄@Cu₁₇₇Se displayed excellent photostability after five heating and cooling cycles under NIR-II photoirradiation (Figure S7, Supporting Information). Noted that Fe₃O₄@Cu_{1.77}Se remained in the original morphology and absorbance after photoirradiation, suggesting that Fe₃O₄@Cu_{1.77}Se could not be collapsed into small nanoparticles with elevated temperature (Figures S8, S9, Supporting Information).

To enhance the in vivo stability and biocompatibility of Fe₃O₄@Cu_{1.77}Se, PEG was afterward conjugated on their surface. Dynamic light scattering (DLS) showed Fe₃O₄@Cu₁₇₇Se -PEG had the hydrodynamic size of 195 nm and good colloidal stability up to 7 days in water (Figure S10, Supporting Information). Zeta potentials of Fe₃O₄, Cu_{1.77}Se, Fe₃O₄@Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se-PEG were 8, -17, -8.5, and -12 mV, respectively (Figure S11, Supporting Information). The strong band at 588 cm⁻¹ of Fe₃O₄ in Fourier transform IR (FTIR) spectroscopy was due to Fe-O vibration, which was weakened and shifted to 581 cm⁻¹ for Fe₃O₄@Cu_{1.77}Se. The peaks at 1053 (C–O groups), 1632 (characteristic C=C stretching groups), 2845 and 2924 cm⁻¹ (asymmetric and symmetric stretching vibrations of -CH2groups, respectively), and 3437 cm^{-1} (–OH groups), proving the existence of gelatin, were all weakened after Cu177Se conjugation (Figure S12, Supporting Information). Thermogravimetric analysis (TGA) further indicated the content of gelatin, MSA, and PEG in Fe₃O₄@Cu₁₇₇Se was 6.52%, 12.74%, and 4.83%, respectively (Figure S13, Supporting Information).

Next, MMP-2 activatable, NIR-II photothermal-enhanced catalytic activities, and glutathione (GSH) depletion capabilities of Fe₃O₄@Cu_{1.77}Se were examined. Fe₃O₄@Cu_{1.77}Se was pretreated with or without MMP-2 for different time intervals, and generation of hydroxyl radicals (•OH) and depletion of GSH were then evaluated by monitoring the decreased absorption of methylene blue (MB) and 5,5'-dithiobis (2-nitrobenzoic acid) (DTNB), respectively (**Figure 2**a). MB can be degraded by •OH to form MB-OH with low absorbance ranging from 600 to 750 nm in UV–vis spectroscopy, while DTNB can be reduced by

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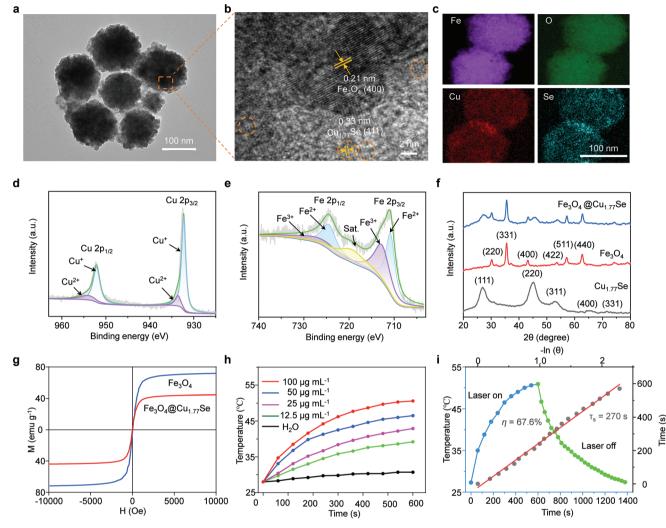


Figure 1. Synthesis and characterization of $Fe_3O_4@Cu_{1.77}$ Se nanozymes. a) TEM, b) HRTEM images, and c) elemental mapping of $Fe_3O_4@Cu_{1.77}$ Se. XPS spectra of d) Cu and (e) Fe in $Fe_3O_4@Cu_{1.77}$ Se. f) Powder XRD patterns of Fe_3O_4 (JCPDS 19-0629), $Cu_{1.77}$ Se (JCPDS 06-0680), and $Fe_3O_4@Cu_{1.77}$ Se. g) Room temperature magnetization curve of Fe_3O_4 and $Fe_3O_4@Cu_{1.77}$ Se. h) Photothermal curves of $Fe_3O_4@Cu_{1.77}$ Se with different concentrations and DI water upon second NIR photoirradiation (1064 nm, 0.75 W cm⁻²). i) Photothermal profile of $Fe_3O_4@Cu_{1.77}$ Se (50 µg mL⁻¹) under second NIR photoirradiation (1064 nm, 0.75 W cm⁻²). ii) Photothermal room temperature, and linear time data versus $ln(\theta)$ acquired from the cooling period of panel.

GSH to form 2-nitro-5-thiobenzoic acid with maximal absorption at 412 nm. As shown in Figure 2b, in the presence of H_2O_2 (5 mm, mimicking the TME), the bleach of MB absorbance from Fe₃O₄@Cu_{1.77}Se pretreated with MMP-2 for 48 h was the fastest, followed by 24 and 0 h at each tested time points, whereas it remained unchanged in non-MMP-2 treated groups. This indicated that the presence of MMP-2 accelerated the decomposition of H2O2 by Fe3O4@Cu177Se to generate •OH in a time-dependent manner. At 10 min, the MB absorbance (A_{10}/A_0) at 48 h was 1.3-, 1.9- and 1.9-fold lower than those at 24, 0 h, and non MMP-2 pretreated groups, respectively (Figure S14, Supporting Information). Subsequently, the catalytic activities of Fe₃O₄@Cu_{1.77}Se were compared to those of Fe₃O₄ and Cu_{1.77}Se with or without pretreatment of MMP-2 for 48 h, respectively (Figure 2c). Fe₃O₄@Cu₁₇₇Se exhibited the fastest MB degradation capabilities, followed by Cu1.77Se and Fe₃O₄ at each test time point in non-MMP-2 pretreated groups.

The higher catalytic activities of $Cu_{1.77}$ Se compared to Fe_3O_4 are due to higher Fenton-like reaction kinetics of Cu^+ ions compared to Fe^{2+} ions. The catalytic ability of $Fe_3O_4@Cu_{1.77}$ Se was further enhanced upon MMP-2 pretreatment. Specifically, at 10 min, MB bleach of $Fe_3O_4@Cu_{1.77}$ Se was 66.8%, 1.9-, 2.1-, and 1.7-fold lower than that in $Fe_3O_4@Cu_{1.77}$ Se, Fe_3O_4 +MMP-2, and $Cu_{1.77}$ Se+MMP-2 groups, respectively. Notably, upon combinational 10 min photoirradiation, MB bleaching by $Fe_3O_4@Cu_{1.77}$ Se with MMP-2 pretreatment was decreased from 66.8% to 44.5%, confirming that NIR-II photothermal effect significantly accelerated the Fenton reaction rate.

The catalytic efficacy was then evaluated by incubating Fe₃O₄@ Cu_{1.77}Se with different concentrations of H₂O₂ in the presence of MMP-2 at 0 and 48 h, respectively. $K_{\rm m}$ of Fe₃O₄@Cu_{1.77}Se incubated with MMP-2 for 0 and 48 h were determined to be 11.46 and 9.77 mM, respectively. Meanwhile, $V_{\rm m}$ of Fe₃O₄@Cu_{1.77}Se incubated with MMP-2 for 0 and 48 h were 3.57 × 10⁻⁸ and

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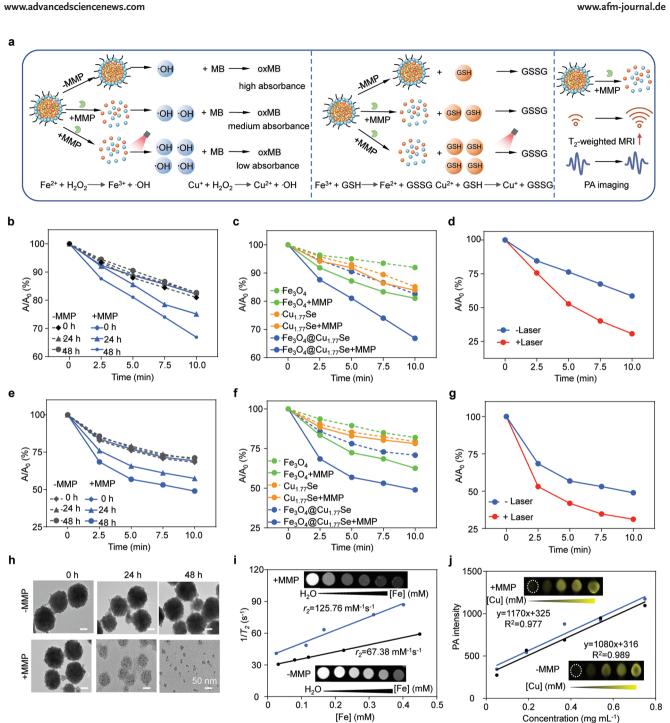


Figure 2. In vitro MMP-2 activatable, second-photothermal boosted catalytic behavior, GSH depletion, and imaging of $Fe_3O_4@Cu_{1.77}Se$ upon treatments. a) Diagram for MMP-2 activatable and NIR-II photoirradiation boosted •OH generation, GSH depletion, and MR/PA imaging of $Fe_3O_4@Cu_{1.77}Se$. b) MB degradation of $Fe_3O_4@Cu_{1.77}Se$ (200 µg mL⁻¹) upon MMP-2 pretreatment for 0, 24, and 48 h. c) MB degradation of $Fe_3O_4@Cu_{1.77}Se$ (200 µg mL⁻¹), vith or without MMP-2 pretreatment for 48 h. d) MB degradation of MMP-2 pretreated $Fe_3O_4@Cu_{1.77}Se$ with or without photoirradiation (1064 nm, 0.75 W cm⁻²) for 10 min. e) GSH depletion of $Fe_3O_4@Cu_{1.77}Se$ (200 µg mL⁻¹) with or without MMP-2 pretreatment for 0, 24, and 48 h. f) GSH depletion of $Fe_3O_4@Cu_{1.77}Se$ (165 µg mL⁻¹), and $Fe_3O_4@Cu_{1.77}Se$ (200 µg mL⁻¹) with or without MMP-2 pretreatment for 0, 24, and 48 h. f) GSH depletion of $Fe_3O_4@Cu_{1.77}Se$ (165 µg mL⁻¹), and $Fe_3O_4@Cu_{1.77}Se$ (200 µg mL⁻¹) with or without MMP-2 pretreatment for 0, 24, and 48 h. f) GSH depletion of $Fe_3O_4@Cu_{1.77}Se$ (165 µg mL⁻¹), and $Fe_3O_4@Cu_{1.77}Se$ (200 µg mL⁻¹) with or without MMP-2 pretreatment. g) GSH depletion of MMP-2 pretreated $Fe_3O_4@Cu_{1.77}Se$ with or without photoirradiation (1064 nm, 0.75 W cm⁻²). h) TEM images of $Fe_3O_4@Cu_{1.77}Se$ with or without MMP-2 pretreatment for 0, 24, and 48 h. i) T₂-weighted MR and j) PA imaging of $Fe_3O_4@Cu_{1.77}Se$ with or without MMP-2 pretreatment for 48 h.

 4.55×10^{-8} M s⁻¹ respectively, indicating stronger catalytic ability of Fe₃O₄@Cu_{1.77}Se after MMP-2 treatment (Figure S15, Supporting Information).

Consistently, GSH depletion abilities of Fe $_3O_4@Cu_{1.77}$ Se was also in MMP-2 activatable, photothermal-enhanced manner. Specifically, the characteristic peak of DTNB at 420 nm

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decreased fastest in Fe₃O₄@Cu_{1.77}Se treated with MMP-2 for 48 h at all the time points. At 10 min, GSH depletion in Fe₃O₄@Cu_{1.77}Se treated with MMP-2 was increased by 1.4and 2.3-fold compared to Fe₃O₄ and Cu_{1.77}Se, respectively (Figure 2f). Combination with photoirradiation at 1064 nm further decreased the DTNB absorbance to 31.2%, 1.2-fold lower than that without photoirradiation (Figure 2g).

Considering that gelatin, the template for the assembly of Fe₃O₄@Cu_{1.77}Se, could be degraded by matrix metalloproteinases, such as MMP-2 or MMP-9, Fe₃O₄@Cu_{1.77}Se in the presence or the absence MMP-2 treatments was collected for both TEM and DLS analysis. In the presence of MMP-2, Fe₃O₄@Cu_{1.77}Se disintegrated gradually with decreased size from 124.2 ± 3.0 to 14.6 ± 6.9 nm over time till 48 h (Figure 2h; Figure S16, Supporting Information). In contrast, in the absence of MMP-2, the size and morphology of Fe₃O₄@Cu_{1.77}Se remained unchanged. These data suggested that MMP-2 triggered the disassembly of Fe₃O₄@Cu_{1.77}Se to generate higher concentrations of reactive iron pools in solution, leading to increased •OH generation and GSH depletion.

Prior to assessing the in vitro therapeutic efficacy of Fe₃O₄@ Cu_{1.77}Se towards 4T1 cells, the cellular uptake and intracellular localization of Fe₃O₄@ Cu_{1.77}Se were examined (Figures S17 and S18, Supporting Information). Fe₃O₄@ Cu_{1.77}Se was conjugated with Cy5, and the fluorescence of Cy5, LysoTracker Green, and Hoechst were utilized to indicate the Fe₃O₄@ Cu_{1.77}Se, lysosome, and nuclei, respectively. Increased cellular uptake of Fe₃O₄@Cu_{1.77}Se was observed with increasing incubation time and reached the maximum at 6 h. Moreover, the red fluorescence of Cy5 overlapped well with the green fluorescence from LysoTracker Green, suggesting the localization of Fe₃O₄@Cu_{1.77}Se in the lysosomes.

Afterward, in vitro therapeutic efficacy of Fe₃O₄@Cu₁₇₇Se was assessed. In the absence of H₂O₂, Fe₃O₄@Cu₁₇₇Se had negligible toxicity to 4T1 cells even with a concentration up to 100 mg mL⁻¹, indicating its good biocompatibility (Figure S19, Supporting Information). However, in the presence of H₂O₂ (5 mm), the viability decreased to 48.4% in Fe₃O₄@Cu_{1.77}Setreated cells, which was 2.3- and 1.5-fold higher than that in Fe₃O₄- and Cu₁₇₇Se-treated groups, respectively. Upon treated with MMP-2, cell viabilities were further decreased to 37.5% and 63.2% in the Fe₃O₄@Cu_{1.77}Se- and Fe₃O₄-treated groups, respectively, indicating the enhanced cytotoxicity from the disassembly of Fe₃O₄ and Fe₃O₄@Cu_{1.77}Se in the presence of MMP-2 (Figure 3a). With combinational photoirradiation at 1064 nm for 3 min, the cell viability further decreased to 19.9%, which was 1.8- and 1.2-fold lower than those in Fe₃O₄- and Cu_{1.77}Setreated cells, respectively, at the same condition.

Such excellent in vitro therapeutic efficacy can be ascribed to the synergistic enhanced •OH production and GSH-depletion of Fe₃O₄@Cu_{1.77}Se upon combinational treatments with MMP-2 and NIR-II photoirradiation, as evident in Figures 3b,c,e. Significant green fluorescence of DCFH-DA (2',7'-dichlorofluorescein diacetate, a fluorescent turn-on indicator for ROS) was only observed in Fe₃O₄@Cu_{1.77}Se-treated cells with MMP-2, while it was weak in both Fe₃O₄- and Cu_{1.77}Se-treated cells in the presence or absence of MMP-2. With combinational NIR-II photoirradiation, Fe₃O₄@Cu_{1.77}Se-treated cells showed a higher fluorescence compared with that in the absence of photoirra-

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diation, indicating that photothermal treatment significantly facilitated the generation of •OH from Fe₃O₄@Cu_{1.77}Se. Meanwhile, DTNB assay indicated that Fe₃O₄@Cu_{1.77}Se+MMP-2+photoirradiation-treated cells reduced GSH content to 20.9%, the most significant, while Fe₃O₄@Cu_{1.77}Se- and Fe₃O₄@ Cu_{1.77}Se+MMP-2-treated groups depleted GSH to 50.2% and 35.2%, respectively (Figure 3c). Fe₃O₄- or Cu_{1.77}Se-treated cells, regardless of treatments with MMP-2 or photoirradiation, exhibited much weaker GSH depletion capabilities than that in Fe₃O₄@Cu_{1.77}Se-treated cells, at the same conditions.

It has been reported that overproduction of •OH and depletion of GSH to inactivate the glutathione peroxidase 4 (GPX4) can lead to the excess accumulation of lipid peroxidation (LPO) inside cells, which is a critical biomarker of ferroptosis.^[18] Therefore, the intracellular expression of LPO in 4T1 cells after treatments was then evaluated using BODIPY665/676-C11 (a LPO sensor) by confocal laser scanning microscopy (CLSM) (Figure 3d,f). Highest fluorescence was observed in Fe₃O₄@ Cu1777Se-treated cells than that in Fe3O4- and Cu177Se-treated cells, suggesting the greatest LPO generation. The signal was further enhanced from 29.75% to 34.9% when Fe₃O₄@Cu₁₇₇Setreated cells were incubated with MMP-2. Combinational photoirradiation at 1064 nm for 4 min further boosted the fluorescence to 44.6%, which was 1.6- and 1.3-fold higher than those in Fe₃O₄- and Cu₁₇₇Se-treated cells, respectively, at the same condition.

To further verify the ferroptosis-inducing capability of Fe₃O₄@Cu₁₇₇Se, western blot was performed to examine the expression of ferroptosis-related biomarkers, including ferritin, ferroportin-1 (FPN-1), long-chain-fatty-acid-CoA ligase 4 (ACSL4), and GPX4 (Figure 3h). Ferritin is the major intracellular iron storage protein.^[19] FPN-1 is the sole iron exporter, of which the downregulation has been indicated to increase intracellular iron retention.^[20] ACSL4 is an isozyme that is responsible for the esterification of CoA to free fatty acids and the formation of acyl-CoA activates the corresponding fatty acids for LPO.^[21] High ACSL4 expression renders cells more sensitive to ferroptosis by preferentially catalyzing arachidonic acid (AA), and shaping cellular lipid composition. As shown in Figure 3h, the most significantly downregulated expressions of ferritin, FPN-1, ACSL4, and GPX4 were all observed in Fe₃O₄@ Cu₁₇₇Se+MMP-2+photolirradiation-treated cells, compared with those in groups of PBS, Fe₃O₄@Cu_{1.77}Se+MMP-2, Fe₃O₄@ Cu177Se+photoirradiation. Meanwhile, the highest expression of caspase-3 (cas-3), the key enzyme in cell apoptosis, was also observed in Fe₃O₄@Cu_{1.77}Se+MMP-2+photoirradiation-treated cells, compared with that in groups of Fe₃O₄@Cu_{1.77}Se+MMP-2 and Fe₃O₄@Cu₁₇₇Se+photoirradiation, suggesting the occurrence of cas-3 dependent apoptosis. This implies that the degradation of ferritin and FPN-1 by Fe₃O₄@Cu_{1.77}Se upon combinational MMP-2 and photoirradiation treatments synergistically generated more reactive iron pools within cells to induce the imbalanced oxidative stress for enhanced ferroptosis/apoptosis against cancer cells.

The ferroptotic/apoptotic cancer cells induced by $Fe_3O_4@Cu_{1.77}Se$ upon combinational MMP-2 and photoirradiation treatments also elicited the immune response by generation of damage-associated molecular patterns (DAMPs), including calreticulin (CRT), adenosine triphosphate (ATP),

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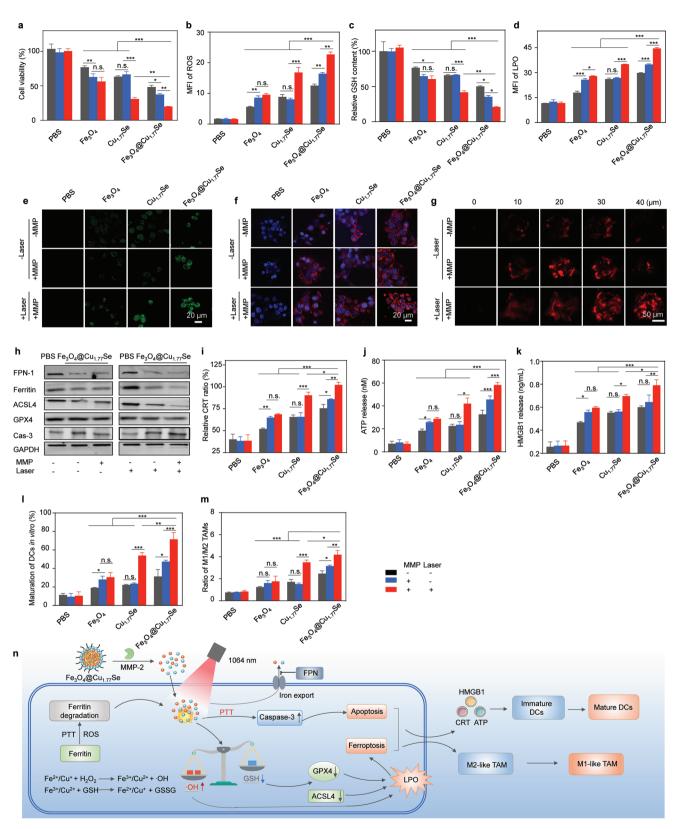


Figure 3. In vitro photothermal ferroptosis anti-tumor therapeutic efficacy evaluation of $Fe_3O_4@Cu_{1.77}Se$ upon treatments. a) Cell viability of 4T1 cells incubated with Fe_3O_4 , $Cu_{1.77}Se$, and $Fe_3O_4@Cu_{1.77}Se$ at the concentration of 50 µg mL⁻¹ with or without MMP, with MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²) in the presence of H_2O_2 . b) Fluorescence quantification of ROS in 4T1 cells after incubating with Fe_3O_4 , $Cu_{1.77}Se$, and $Fe_3O_4@Cu_{1.77}Se$ at concentration of 50 µg mL⁻¹ with or without MMP, with MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²) in the presence of H_2O_2 . b) Fluorescence quantification of ROS in 4T1 cells after incubating with Fe_3O_4 , $Cu_{1.77}Se$, and $Fe_3O_4@Cu_{1.77}Se$ at concentration of 50 µg mL⁻¹ with or without MMP, with MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). c) Relative GSH content in 4T1 cells incubated with Fe_3O_4 , $Cu_{1.77}Se$, and $Fe_3O_4@Cu_{1.77}Se$ at concentration of 50 µg mL⁻¹ with or without MMP, with MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). c) Relative GSH content in 4T1 cells incubated with Fe_3O_4 , $Cu_{1.77}Se$, and $Fe_3O_4@Cu_{1.77}Se$ at concentration of 50 µg mL⁻¹ with or without MMP, with MMP and

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and high mobility group box 1 (HMGB1), as evident in Figures 3i-k. Elevated level of CRT in the Fe₃O₄@Cu₁₇₇Se-treated cells (75.4%) was observed, as compared to that in Fe_3O_4 (51.9%) and Cu_{1.77}Se (65.54%)-treated cells. With MMP-2 treatment, the signals were enhanced to 85.9% and 64.8% in Fe₃O₄@Cu₁₇₇Se and Fe₃O₄-treated cells, respectively, whereas similar CRT level was observed in Cu177Se-treated cells (Figure S20, Supporting Information). Combinational photoirradiation at 1064 nm for 3 min boosted the CRT content in Fe₃O₄@Cu₁₇₇Se-treated cells, which was 1.5- and 1.1-fold higher than those in Fe₃O₄ and Cu177Se-treated cells, respectively, at the same condition. Similar trends were also observed for the release of ATP and HMGB1 in Fe₃O₄@Cu_{1.77}Se+MMP-2+photoirradiation-treated cells (Figure 3j,k). These results verified that Fe₃O₄@Cu₁₇₇Se by combinational MMP-2 and photoirradiation treatments triggered efficient generation of immunogenic cell death (ICD).

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Since the translocation of CRT in endoplasmic reticulum to cell surface, release of ATP, and exposure of HMGB1 on the surface of apoptotic cells can facilitate the uptake, processing, and presentation of tumor antigens by dendritic cells (DCs), the capability of Fe₃O₄@Cu₁₇₇Se to promote DCs maturation was next assessed.^[22] The matured ratio of DCs (CD80⁺CD86⁺) for Fe₃O₄@Cu₁₇₇Se+MMP-2+photoirradiation was 71.6%, which was 2.3- and 1.5-fold higher than that in groups of Fe₃O₄@Cu_{1.77}Se and Fe₃O₄@Cu₁₇₇Se+MMP-2, respectively (Figure S21, Supporting Information). By contrast, Fe₃O₄- and Cu_{1.77}Se-treated groups showed obviously decreased DCs maturation ratio. Noted that with photoirradiation, nearly one-fold enhanced maturation ratio of DCs by Cu177Se was observed, indicating that photothermal treatment elevated the levels of ICD to facilitate the DCs maturation. Along with DCs maturation, the polarization of M2-like macrophages to M1 phenotype by Fe₃O₄@Cu_{1.77}Se was evaluated in IL-4-conditioned RAW 264.7 cells (Figure 3m; Figure S22, Supporting Information). Compared with PBS, Fe₃O₄, and Cu_{1.77}Se, Fe₃O₄@Cu_{1.77}Se exhibited significantly enhanced expression of CD80⁺ and CD86⁺ (M1-related markers), while decreased expression of CD206+ (M2-related marker), suggesting the excellent M2 macrophage repolarization activity of Fe₃O₄@Cu₁₇₇Se. Together, these results verified that Fe₃O₄@Cu_{1.77}Se by MMP-2 and photoirradiation not only triggered efficient generation of ICD to induce the maturation of DCs, but also reprogramed M2 phenotype to M1 macrophages in vitro. The molecular ferroptosis-based therapeutic mechanism and immune stimulation of Fe₃O₄@Cu_{1.77}Se-mediated cancer therapy in vitro were summarized in Figure 3n.

Of note, the disassembled $Fe_3O_4 @\,Cu_{1.77}Se$ upon MMP-2 and photoirradiation treatment exhibited deeper penetration

depth in 4T1 multicellular spheroids (MCS), at each tested depth, compared to those without treatment and MMP-2 only (Figure 3g). At the depth of 30 μ m, the mean fluorescence intensity of Fe₃O₄@Cu_{1.77}Se after photoirradiation was 4.5- and 2.2-fold higher than that without treatments and MMP-2 only (Figure S23, Supporting Information).

After verifying the excellent therapeutic performance in vitro, the feasibility of $Fe_3O_4@Cu_{1.77}$ Se for cancer theranostics in vivo was examined. First, the potential of $Fe_3O_4@Cu_{1.77}$ Se as MMP-2 activatable MRI contrast agents in vivo was evaluated by intratumoral (i.t) injection. The tumor models were established by subcutaneous injection of 4T1 cells into both flanks of each mouse, followed by intratumoral injection of $Fe_3O_4@Cu_{1.77}$ Se into the right tumor site of the mice. Afterward, both sides were imaged at different time points. Enhanced MRI signals overtime were only observed in the right tumor site of mice treated with $Fe_3O_4@Cu_{1.77}$ Se, whereas signal changes were negligible in the nontreated left tumor tissues, validating the MMP-2 activatable MRI contrast enhancement ability of $Fe_3O_4@Cu_{1.77}$ Se for tumor-specific imaging and theranostics (**Figures 4**a,b).

Subsequently, in vivo therapeutic performance of Fe_3O_4 Cu_{1.77}Se was evaluated. 4T1 tumor-bearing BALB/c mice models were established by subcutaneous injection of 4T1 cells into both flanks of each mouse as the primary and distant tumors. After the primary tumor volume reached 100 mm³, mice were intravenously injected with PBS, Fe₃O₄, Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se, followed by local NIR-II photoirradiation (Figure 4c). The growths of both primary and distant tumors and lung metastasis were monitored after treatments. To verify the optimal time-point for photoirradiation, MR and PA imaging were conducted on the 4T1 tumor-bearing mice. Noted that an external magnetic field was applied to the primary tumor of mice to enhance the accumulation of nanoparticles. Both MRI and PA signals in tumor tissues after Fe₃O₄@Cu₁₇₇Se injection gradually increased over time and reached the plateau at 12 h post-injection time (Figures 4d-g). Therefore, therapeutic treatment was conducted by photoirradiation of the primary tumor with 1064 nm laser at 12 h post-injection of Fe₃O₄@Cu_{1.77}Se. The tumors of Fe₃O₄@Cu₁₇₇Se-injected mice exhibited the highest temperature increase, followed by Cu_{1.77}Se and Fe₃O₄ after 10 min photoirradiation, which can be attributed to the higher accumulation of Fe₃O₄@Cu_{1.77}Se in the tumor due to the external magnetic field (Figure 4h; Figure S24, Supporting Information). After photoirradiation, the primary tumors in Fe₃O₄@Cu₁₇₇Se-injected mice were almost inhibited, while the growth of primary tumors in Fe₃O₄ and Cu_{1.77}Se was inhibited by 1.5- and 1.3-fold compared to those in the control mice

photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). d) Quantification of LPO in 4T1 cells incubated with Fe₃O₄, Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se at concentration of 50 μ g mL⁻¹ with or without MMP, with MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). e) Confocal fluorescence images of ROS production and f) LPO content in 4T1 cells after incubating with Fe₃O₄, Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se at concentration of 50 μ g mL⁻¹ with or without MMP, with MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). ROS were indicated by DCFH-DA (green fluorescence), LPO was stained with the red-fluorescent probe BODIPY 665/676, and nuclei were stained with DAPI. g) Confocal fluorescence images of 4T1 MCSs incubated with Fe₃O₄, Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se at concentration of 50 μ g mL⁻¹ with or without MMP, with MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). h) Western blots analysis of expression of ferroptosis and apoptosis-related proteins in 4T1 cells after incubated with Fe₃O₄, Cu_{1.77}Se at concentration of 50 μ g mL⁻¹ with or without MMP, with MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). h) Western blots analysis of expression of ferroptosis and apoptosis-related proteins in 4T1 cells after incubated with Fe₃O₄, Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se at concentration of 50 μ g mL⁻¹ with or without MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). h) Western blots analysis of expression of ferroptosis and apoptosis-related proteins in 4T1 cells after incubated with Fe₃O₄, Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se at concentration of 50 μ g mL⁻¹ with or without MMP and photoirradiation for 3 min (1064 nm, 0.75 W cm⁻²). i) Flow cytometry quantification of CRT on the surface of 4T1 breast tumor cells. j) ATP secretion from 4T1 breast tumor cells was measured by ATP assay kit. k) HMGB1 released by 4T1 breast tumor cells was quantified via ELISA kit. l) Flow cytometry quantification of DCs maturation after inc



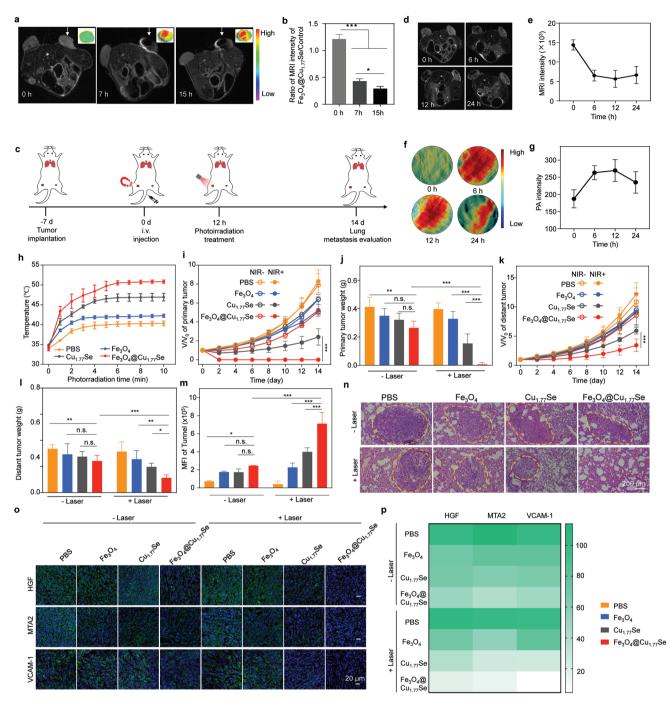


Figure 4. In vivo imaging-guided therapeutic evaluation of $Fe_3O_4@Cu_{1.77}Se$ upon treatments. a) MR imaging and b) the ratio of MRI signal intensity of $Fe_3O_4@Cu_{1.77}Se$ vs PBS after i.t. injection. The left tumors were injected with PBS and the right tumors were injected with $Fe_3O_4@Cu_{1.77}Se$ (2 mg mL⁻¹, 20 µL). c) Schedule of tumor implantation and therapeutic evaluation of $Fe_3O_4@Cu_{1.77}Se$ upon treatments. d) MR imaging and e) quantification of 4TI tumors after i.v. injection of $Fe_3O_4@Cu_{1.77}Se$ with magnet treatment. f) PA imaging and g) quantification of 4TI tumors after i.v. injection of $Fe_3O_4@Cu_{1.77}Se$, and hotoirradiation (0.75 W cm⁻², 10 min) at 12 h after i.v. injection of PBS, Fe_3O_4 , $Cu_{1.77}Se$, and $Fe_3O_4@Cu_{1.77}Se$, respectively. Tumor growth curves of i) primary and k) distant 4TI tumors after treatments. Weight of j) primary and l) distant 4TI tumors from scarificed mice after treatments on day 14. m) Quantification of Tunnel imaging in 4TI tumors after treatments. n) H&E staining of lungs after treatments. cell nuclei were stained by DAPI with blue fluorescence. HGF, MTA2, and VCAM-1 were stained by respective antibodies with green fluorescence. *p < 0.05, **p < 0.01, and ***p < 0.001.

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(Figures 4i,j; Figure S25, Supporting Information). In distant tumors, the inhibition rate in Fe₃O₄@Cu_{1.77}Se-injected mice with photoirradiation was 3.9-, 3.0-, and 1.8-fold lower compared to PBS, Fe₃O₄ and Cu_{1.77}Se, respectively (Figures 4k,l).

Moreover, fewer metastatic nodules were observed in the lungs of Fe₃O₄@Cu₁₇₇Se-injected mice with photoirradiation, while obvious pulmonary tumor metastases were observed for mice in other treatment groups (Figure 4p; Figures S26, S27, Supporting Information). Quantitative analysis showed that the number of metastatic nodules in the lungs of Fe₃O₄@Cu₁₇₇Seinjected mice with photoirradiation was 3.2-, 2.7-, and 1.5-fold lower than those in the Fe₃O₄@Cu_{1.77}Se-injected mice without photoirradiation, Fe₃O₄-, and Cu_{1.77}Se-injected mice with photoirradiation, respectively. Of note, no visible nucleus and cytoplasm alterations or abnormalities in major organs, including heart, liver, spleen, and kidney from mice were observed in the histological examination by hematoxylin and eosin (H&E) stain on day 14 (Figure S28, Supporting Information). No weight loss in the mice was noticed by monitoring the body weight for 14 days after treatment (Figure S29, Supporting Information).

To understand how Fe3O4@Cu177Se inhibited the metastasis, the expressions of metastasis-related proteins including hepatocyte growth factor (HGF), metastasis-associated protein 2 (MTA2), and vascular cell adhesion molecule-1 (VCAM-1) in tumor tissues after treatments were investigated by immunofluorescence staining (Figures 40,p). Weak green fluorescence of HGF, MTA2, and VCAM-1 staining was observed in Fe₃O₄@ Cu_{1.77}Se-injected tumors with photoirradiation, indicating low expressions of these metastasis-related proteins. By contrast, intense green fluorescence signals were observed in the tumors for other treatment groups. Quantitative studies indicated that the expressions of HGF, MTA2, and VCAM-1 were reduced to 17.8%, 9.7%, and 4.5% in Fe₃O₄@Cu₁₇₇Se-injected tumors with photoirradiation, respectively, compared with 67.6%, 60.2%, and 76.3% in Fe₃O₄, 65.6%, 58.7%, and 71.5% in Cu₁₇₇Se and 96.8%, 96.8%, and 95.0% in PBS-injected tumors, respectively. Collectively, these data suggested the synergistic antitumor effect of $Fe_3O_4@Cu_{1.77}Se$ for cancer therapy.

To gain more insight into the in vivo synergistic therapeutic mechanism of Fe₃O₄@Cu_{1.77}Se, generation of ROS, and expression of ferroptosis/apoptosis proteins, including GPX4, ACSL-4, LPO as well as caspase-3, in tumor tissues after 24 h postinjection of PBS, Fe₃O₄, Cu_{1.77}Se, and Fe₃O₄@Cu_{1.77}Se were collected for further immunofluorescence staining (Figures 5a-f). Overall, Fe₃O₄@Cu_{1.77}Se-injected tumors with photoirradiation displayed the highest ROS generation and ferroptosis/ apoptosis-inducing capabilities than those in other treatments. For instance, with 10 min photoirradiation, the fluorescence of DCFH-DA in Fe₃O₄@Cu_{1.77}Se-injected tumors was enhanced from 55.4% to 99.1%, compared to that in PBS (from 7.2% to 7.5%), Fe₃O₄ (from 24.8% to 42.2%), and Cu_{1.77}Se (from 39.0% to 74.2%)-injected tumors. The expressions of GPX-4 and ACSL4 in Fe₃O₄@Cu_{1.77}Se-treated tumor with photoirradiation were decreased from 53.9% to 11.5%, and 42.5% to 10.2%, respectively, which subsequently induced significantly stronger fluorescence of LPO. Similarly, caspase-3 expression in Fe₃O₄@ Cu177Se-treated tumors with photoirradiation was 93.9%, in contrast to those cells treated with PBS (6.3%), Fe₃O₄ (26.9%), or Cu_{1.77}Se (28.1%), respectively.

Subsequently, the generation of HMGB1 and CRT, maturation of DCs in lymph nodes, and activation of immune response in vivo were investigated. HMGB1 in the Fe₃O₄@ Cu1.77Se-injected tumors with photoirradiation was increased by 9.1-, 2.6-, and 1.4-fold compared to PBS, Fe₃O₄, and Cu₁₇₇Se group, respectively, at the same condition (Figure 5g). Similarly, the level of CRT in the Fe3O4@Cu177Se-injected tumor with photoirradiation was 11.4-,2.8-, and 1.4-fold compared to that in PBS-, Fe₃O₄-, and Cu₁₇₇Se-injected groups, respectively (Figure 5h). The efficient generation of ICD triggered by Fe₃O₄@Cu₁₇₇Se in vivo promotes the maturation of DCs, activation of CD8⁺ T cells, and polarization of tumor-associated macrophage. Specifically, with photoirradiation, the populations of matured DCs for Fe₃O₄@Cu₁₇₇Se groups increased by 1.4-, 1.3-, and 1.7-fold compared to Fe₃O₄, Cu₁₇₇Se and control groups, respectively (Figure 5i; Figure S30, Supporting Information). Meanwhile, the highest population of CD8⁺ T cells was also observed in Fe₃O₄@Cu₁₇₇Se-treated tumors with photoirradiation, which was 1.9-, 2.0-, and 2.4-fold higher than those for Fe₃O₄, Cu₁₇₇Se and PBS groups, respectively (Figure 5j; Figure S31, Supporting Information). The polarization of M2 to M1 in $Fe_3O_4@Cu_{1.77}$ Se-treated tumor with photoirradiation increased by 2.2, 1.5-, and 2.7-fold compared to Fe₃O₄, Cu₁₇₇Se and control groups, respectively (Figure 51). By contrast, no noticeable changes in the populations of matured DCs, infiltrative CD8⁺ T cells, and polarization of M2 to M1, were observed for other treatment groups. With photoirradiation, the serum levels of IFN- γ were increased by 1.6- and 1.4-fold in Fe₃O₄@ Cu177Se-injected tumors as compared to Fe3O4- and Cu177Seinjected tumors, respectively. Similarly, with photoirradiation, TNF- α was increased by 1.4-, and 1.2-fold in Fe₃O₄@Cu_{1.77}Seinjected tumors as compared to that in Fe₃O₄- and Cu_{1.77}Seinjected tumors, respectively (Figures 5k,m). The immune stimulation of Fe₃O₄@Cu₁₇₇Se-mediated photothermal ferroptosis cancer therapy in vivo was summarized in Figure 5n. Together, these results provided strong evidence that the Fe₃O₄@Cu₁₇₇Semediated ferroptosis-based second NIR photothermal therapy not only generated strong ICD to induce maturation of DCs that facilitated the activation and infiltration of cytotoxic CD8⁺ T cells into the immunogenetic cold tumor (4T1), but also polarized M2 tumor-associated macrophages to M1 tumor-associated macrophages, thereby leading to the effective inhibition of primary and distant tumors as well as lung metastasis.

3. Conclusions

In summary, we have developed a novel MMP-2 activatable, second NIR photothermal-enhanced theranostic nanozyme based on Fe₃O₄@Cu_{1.77}Se for high-performance cancer therapy. The overexpressed MMP-2 in tumor microenvironment disassembles Fe₃O₄@Cu_{1.77}Se to enhance reactive intratumor ferrous supply for elevated ferrotherapeutic effect and enhanced T₂-weighted MRI contrast. Combinational NIR-II photoirradiation not only triggers effective photothermal therapy, but more importantly, significantly potentiates Fenton reaction to generate substantial hydroxyl radicals and consume GSH to induce more lipid peroxidation for boosted ferroptosis. Such synergism induces immunogenic cell death to enhance the



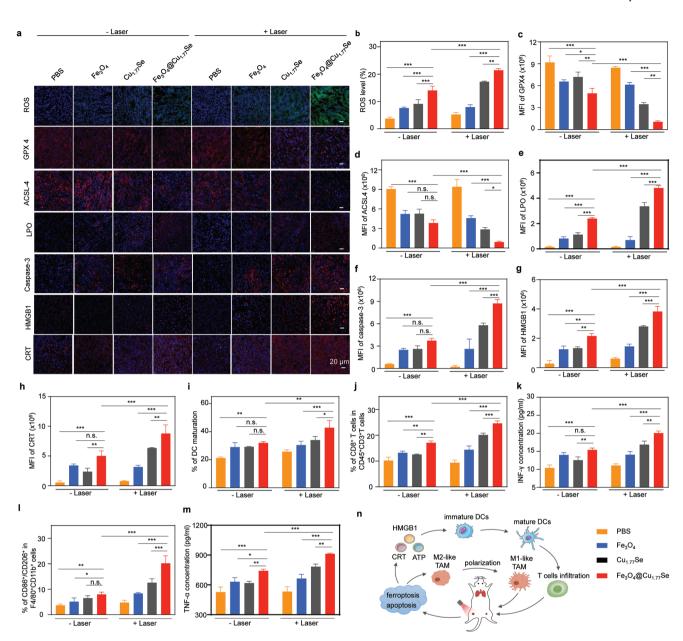


Figure 5. In vivo therapeutic mechanism of Fe₃O₄@Cu_{1.77}Se for photothermal ferroptosis cancer therapy. a) Immunofluorescence images and b–h) quantification of ROS, LPO, GPX-4, ACSL-4, caspase-3, HMGB1, and CRT in 4T1 tumors after treatments. ROS was indicated with green color. LPO, GPX-4, ACSL-4, caspase-3, HMGB1, and CRT were indicated with red color. Cell nuclei were stained by DAPI with blue color. i) Quantification of DCs maturation in lymph nodes on day 3 after treatments (n = 3). j) Quantification of the ratio of CD8⁺T cells in CD45⁺CD3⁺T cells of 4T1 tumors on day 5 after treatments (n = 3). k) Quantitative analysis of IFN- γ in the serum of 4T1 tumor-bearing mice after different treatments on day 14 (n = 3). l) Ratio of M₁/M₂-like TAM in 4T1 tumors on day 5 after treatments (n = 3). m) TNF- α secretion in the serum of 4T1 tumor-bearing mice after treatments on day 14 (n = 3). n) Mechanistic scheme of Fe₃O₄@Cu_{1.77}Se-induced anticancer effect in vivo. *p < 0.05, **p < 0.01, and ***p < 0.001.

cytotoxic T cells recruitment and infiltration into the immunogenetic cold tumor, and also, polarizes M2 tumor-associated macrophages to M1 tumor-associated macrophages. As such, it can efficiently eliminate the primary tumors, and elicits systemic anti-tumor immunity, contributing to regression of nontreated distant tumors and inhibition of lung metastasis. Thus, this study provides a promising activatable nanozyme design strategy for photothermal ferroptosis cancer therapy with high spatial-temporal specificity.

Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

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Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

The data that support the findings of this study are available in the supplementary material of this article.

Keywords

photoacoustic imaging, photothermal ferroptosis cancer therapy, protease-activatable nanozymes, tumor-enhanced magnetic resonance imaging

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- S. J. Dixon, K. M. Lemberg, M. R. Lamprecht, R. Skouta, E. M. Zaitsev, C. E. Gleason, D. N. Patel, A. J. Bauer, A. M. Cantley, W. S. Yang, B. Morrison3rd, B. R. Stockwell, *Cell* **2012**, *149*, 1060.
- [2] V. Trujillo-Alonso, E. C. Pratt, H. Zong, A. Lara-Martinez, C. Kaittanis, M. O. Rabie, V. Longo, M. W. Becker, G. J. Roboz, J. Grimm, M. L. Guzman, *Nat. Nanotechnol.* 2019, *14*, 616.
- [3] H. Liu, R. Jiang, Y. Lu, B. Shan, Y. Wen, M. Li, ACS App. Mater. Interfaces 2022, 14, 28537.
- [4] S. E. Kim, L. Zhang, K. Ma, M. Riegman, F. Chen, I. Ingold, M. Conrad, M. Z. Turker, M. Gao, X. Jiang, S. Monette, M. Pauliah, M. Gonen, P. Zanzonico, T. Quinn, U. Wiesner, M. S. Bradbury, M. Overholtzer, *Nat. Nanotechnol.* **2016**, *11*, 977.
- [5] X. Liu, X. Zhu, X. Qi, X. Meng, K. Xu, Int. J. Nanomed. 2021, 16, 1037.
- [6] C. Zhang, W. Bu, D. Ni, S. Zhang, Q. Li, Z. Yao, J. Zhang, H. Yao, Z. Wang, J. Shi, Angew. Chem., Int. Ed. 2016, 55, 2101.

- [7] a) W. Ni, Y. Li, L. Liang, S. Yang, M. Zhan, C. Lu, L. Lu, L. Wen, J. Biomed. Nanotechnol. 2022, 18, 327; b) H. Chi, G. Zhu, Y. Yin, H. Diao, Z. Liu, S. Sun, Z. Guo, W. Xu, J. Xu, C. Cui, X.-J. Xing, K. Ma, Int. J. Pharm. 2022, 622, 121898; c) X. Tan, J. Huang, Y. Wang, S. He, L. Jia, Y. Zhu, K. Pu, Y. Zhang, X. Yang, Angew. Chem. Int. Ed. 2021, 60, 14051.
- [8] a) S. He, Y. Jiang, J. Li, K. Pu, Angew. Chem. Int. Ed. 2020, 59, 10633;
 b) C. Xu, K. Pu, Chem. Soc. Rev. 2021, 50, 1111; c) S. Dong, Y. Dong, T. Jia, S. Liu, J. Liu, D. Yang, F. He, S. Gai, P. Yang, J. Lin, Adv. Mater. 2020, 32, 2002439.
- [9] J. Zhu, P. Dai, F. Liu, Y. Li, Y. Qin, Q. Yang, R. Tian, A. Fan, S. d. F. Medeiros, Z. Wang, Y. Zhao, *Nano Lett.* **2020**, *20*, 6235.
- [10] a) Z. Yu, W. K. Chan, Y. Zhang, T. T. Y. Tan, *Biomaterials* **2021**, *269*, 120459; b) A. M. Smith, M. C. Mancini, S. Nie, *Nat. Nanotechnol.* **2009**, *4*, 710; c) J. Huang, J. Li, X. Zhang, W. Zhang, Z. Yu, B. Ling, X. Yang, Y. Zhang, *Nano Lett.* **2020**, *20*, 5236.
- [11] J. Jia, G. Liu, W. Xu, X. Tian, S. Li, F. Han, Y. Feng, X. Dong, H. Chen, Angew. Chem., Int. Ed. 2020, 59, 14443.
- [12] B. Tian, S. Liu, L. Feng, S. Liu, S. Gai, Y. Dai, L. Xie, B. Liu, P. Yang, Y. Zhao, Adv. Funct. Mater. 2021, 31, 2100549.
- [13] Z. Xu, N. Rao, C.-Y. Tang, C.-H. Cheng, W.-C. Law, ACS Omega 2019, 4, 14655.
- [14] a) J. Li, X. Yu, Y. Jiang, S. He, Y. Zhang, Y. Luo, K. Pu, Adv. Mater. 2021, 33, 2003458; b) C. Xu, Y. Jiang, Y. Han, K. Pu, R. Zhang, Adv. Mater. 2021, 33, 2008061.
- [15] Y. Jiang, X. Zhao, J. Huang, J. Li, P. K. Upputuri, H. Sun, X. Han, M. Pramanik, Y. Miao, H. Duan, K. Pu, R. Zhang, *Nat. Commun.* 2020, 11, 1857.
- [16] M. Chang, Z. Hou, M. Wang, C. Yang, R. Wang, F. Li, D. Liu, T. Peng, C. Li, J. Lin, Angew. Chem. Int. Ed. 2021, 60, 12971.
- [17] C. Ou, W. Na, W. Ge, H. Huang, F. Gao, L. Zhong, Y. Zhao, X. Dong, Angew. Chem. Int. Ed. 2021, 60, 8157.
- [18] W. S. Yang, R. SriRamaratnam, M. E. Welsch, K. Shimada, R. Skouta, V. S. Viswanathan, J. H. Cheah, P. A. Clemons, A. F. Shamji, C. B. Clish, L. M. Brown, A. W. Girotti, V. W. Cornish, S. L. Schreiber, B. R. Stockwell, *Cell* **2014**, *156*, 317.
- [19] P. Arosio, L. Elia, M. Poli, IUBMB Life 2017, 69, 414.
- [20] I. Yanatori, D. R. Richardson, K. Imada, F. Kishi, J. Biol. Chem. 2016, 291, 17303.
- [21] S. Doll, B. Proneth, Y. Y. Tyurina, E. Panzilius, S. Kobayashi, I. Ingold, M. Irmler, J. Beckers, M. Aichler, A. Walch, H. Prokisch, D. Trümbach, G. Mao, F. Qu, H. Bayir, J. Füllekrug, C. H. Scheel, W. Wurst, J. A. Schick, V. E. Kagan, J. P. F. Angeli, M. Conrad, *Nat. Chem. Biol.* 2017, 13, 91.
- [22] a) D. V. Krysko, A. D. Garg, A. Kaczmarek, O. Krysko, P. Agostinis, P. Vandenabeele, *Nat. Rev. Cancer* **2012**, *12*, 860; b) M. Fan, L. Jia, M. Pang, X. Yang, Y. Yang, S. Kamel Elyzayati, Y. Liao, H. Wang, Y. Zhu, Q. Wang, *Adv. Funct. Mater.* **2021**, *31*, 2010587.